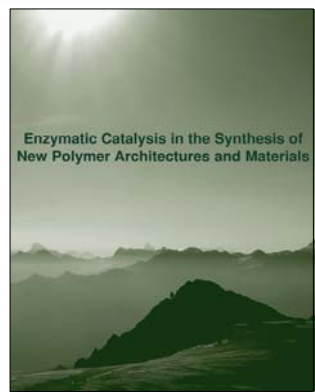


## Enzymatic Catalysis in the Synthesis of New Polymer Architectures and Materials



**Dr. ir. Matthijs de Geus**  
*Laboratory of Polymer Chemistry  
Technische Universiteit Eindhoven*

November 21<sup>st</sup>, 2007  
DPI Annual Meeting 2007



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## DPI interest in biocatalysis



Website Dutch Polymer Institute - 20/11/07

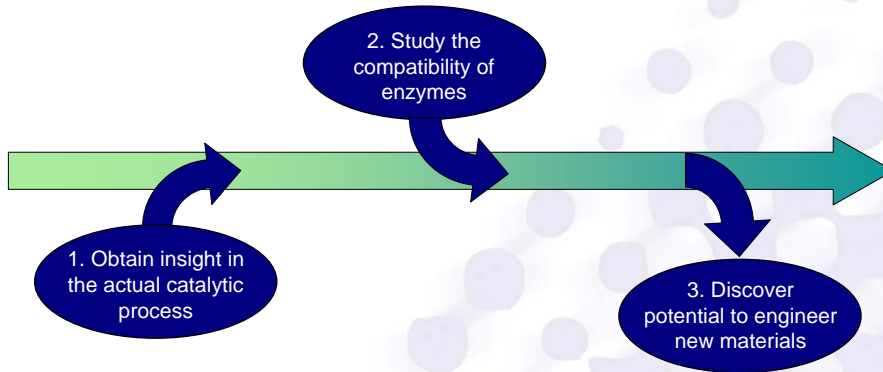
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## Objective Thesis

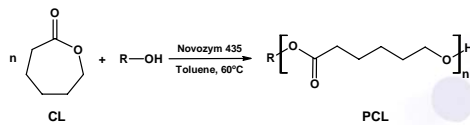
*"Can enzymes offer new perspectives in polymer chemistry?"*



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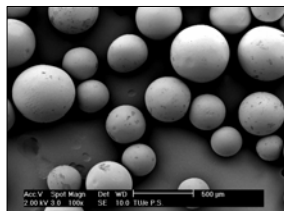
## Concepts

**eROP: Enzymatic ring-opening polymerization**



1993: Knani et al. *J. Pol. Sci. A*, 31, 5, 1221-1232

**Novozym 435**




**Why enzymes?**

- Enzymes are very efficient catalysts
- Enzymes act under mild conditions
- Enzymes are environmentally/biomedically acceptable catalysts
- Enzymes are not bound to their natural role
- Enzymes can catalyze a broad spectrum of reactions
- Enzymes are substrate-selective

These advantages can be utilized to make materials that are not easily available from conventional techniques

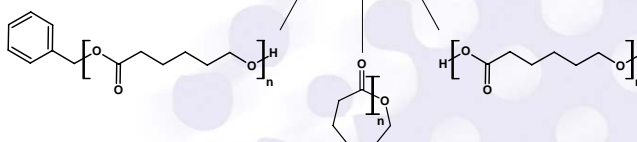
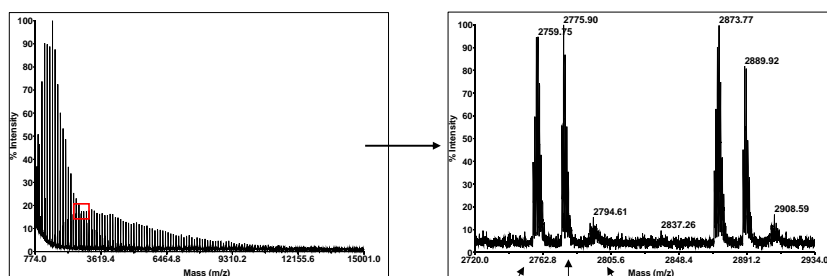
Heel-Hoorn-Denk  
Lectures Polymer '07, 2007

TU/e  **DPI**  
DUTCH POLYMER INSTITUTE

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# 1. Insight in Catalytic Process

## MALDI-TOF-MS



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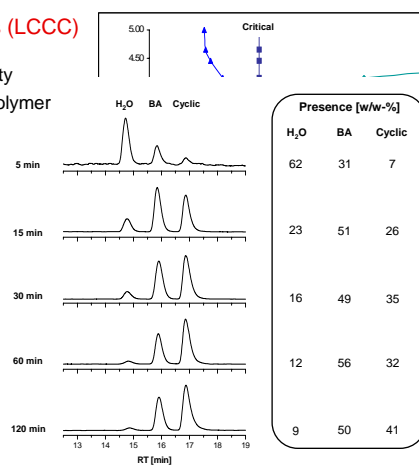
# LCCC

## Liquid Chromatography under Critical Conditions (LCCC)

- > Separation based on difference in end-functionality
- > Only technique to distinguish between different polymer species in a quantitative manner!

### Observations:

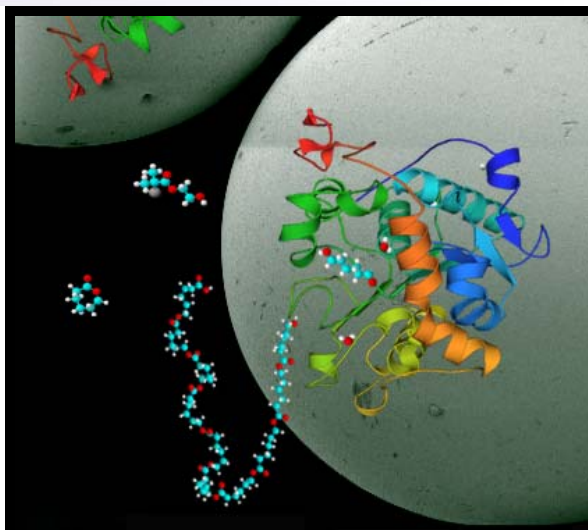
- > Large amounts of polymeric side-products
- > Initial nucleophile is water



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## LCCC

### Liquid Chromatography under Critical Conditions (LCCC)

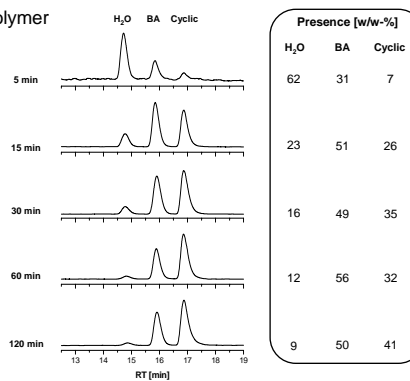
- Separation based on difference in end-functionality
- Only technique to distinguish between different polymer species in a quantitative manner!

#### Observations:

- Large amounts of polymeric side-products
- Initial nucleophile is water

#### Therefore:

- Minimize water concentration
- Optimize initiator performance
- Reduce cycle formation



De Geus et al. *manuscript accepted in Biomacromolecules* 2007

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## Reduce polymer side-products

### Minimize water concentration

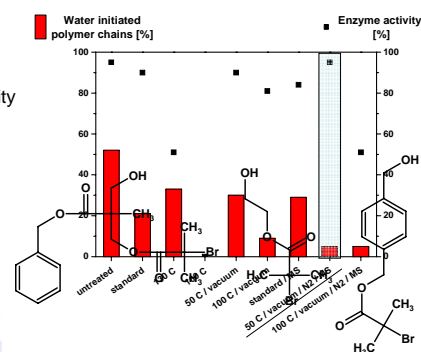
- Enzyme needs water to function
- Develop drying technique that allows enzyme activity with minimal water initiation

### Optimize initiator performance

- Determine criteria for an initiator to compete with other nucleophiles (size, chirality, polarity)

### Reduce cycle formation

- Define appropriate reaction conditions to minimize the formation of cyclic polymer structures  
→ work in bulk



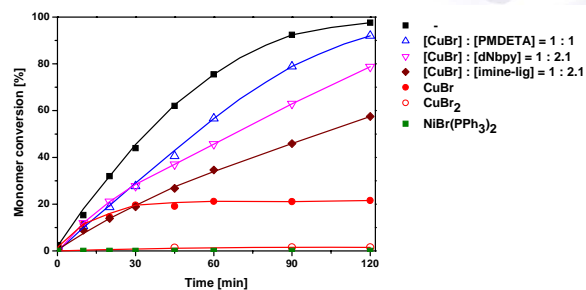
De Geus et al. *Macromolecules* 2005, 38, 4220

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## 2. Compatibility of Enzymes

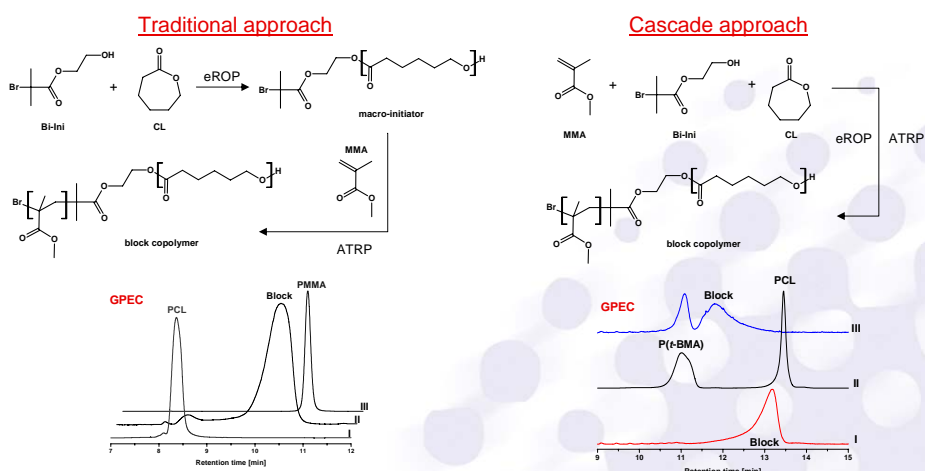
### Model system

- Combination of controlled radical polymerization and eROP
- Check mutual influence:
  - ATRP-system on eROP
  - eROP-system on ATRP



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## Chemoenzymatic Cascade Approach



Duxbury et al. *J. Am. Chem. Soc.* **2005**, 127, 2384  
 De Geus et al. *J. Pol. Sci. A* **2006**, 44, 4290

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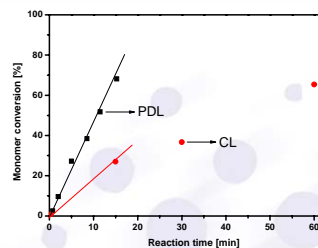
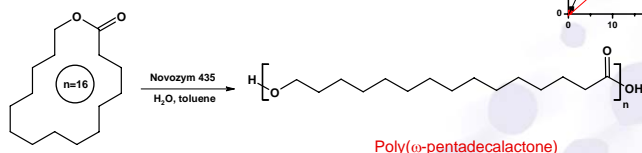


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## 3. Engineer New Materials

### Advantages of enzymes:

- Stereo-, regio- and enantioselectivity
- Different behavior towards large lactones ( $n > 12$ )



- Aliphatic polyester  $\rightarrow$  biocompatibility/biodegradability
- Structural resemblance with linear polyethylene ➔ push  $M_w$  to its limit
- High molecular weight PPDL only accessible with enzymes ➔ Fiber applications!

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## Polypentadecalactone

### High molecular weight PPDL

- Decrease amount of nucleophiles
- Overcome diffusional problems during polymerization (heterogeneous catalyst)
- Molecular analysis at elevated temperature (high crystallinity)

$M_w$ [kg/mol]	$M_w/M_n$ [-]	$T_m$ [°C]	$T_g$ [°C]	$\Delta H_m$ [J/g]	$\chi_c$ [-]
190	3.2	95	-25	138	59

Polymer	$T_g$ [°C]	$T_m$ [°C]	E [MPa]	$\sigma_{break}$ [MPa]	$\sigma_{yield}$ [MPa]	$\epsilon_{yield}$ [%]	$\epsilon_{break}$ [%]
PLLA	57	170-180	1000-3000	n.d.	50	<4	<7
PCL	-60	55-60	400	n.d.	16	7.0	800-1000
<b>PPDL</b>	<b>-25</b>	<b>95</b>	<b>420</b>	<b>38</b>	<b>17.5</b>	<b>15</b>	<b>1200</b>
LLDPE	-60	122-130	250-500	13-27	9-19	-	100-965
LDPE	-25	100-120	200-300	8-30	9-14	-	100-650
HDPE	-125	130-140	1100	21-32	n.d.	-	1200

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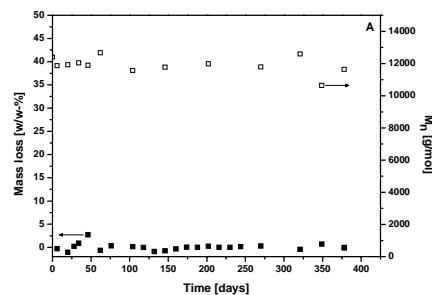
## Future application of PPDL

### Biomedical fiber applications

- Hydrophobic and crystalline polymer material
- Metal-free synthesis
- Can be chemically incorporated with other (hydrophilic) segments:
  - PEG
  - lactones
  - lactide/glycolide

Material could be used as fiber or tape for:

- Suturing
  - Bone repair
  - Tissue engineering
  - Stents
- Continuation of this work in DPI #608



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## Highlights

### Enzymatic polymerization

1. Obtain insight in actual catalytic process
  - ✓ Development of LCCC to quantify different polymer species
  - ✓ Water is initially most active nucleophile in eROP
  - ✓ Optimize reaction conditions to minimize side-products
2. Study the compatibility of enzymes towards other polymerization techniques
  - ✓ Investigation of mutual influence
  - ✓ Combination of eROP and controlled radical polymerization in one pot (cascade)
3. Discover its potential to engineer new materials
  - ✓ Development of high molecular weight PPDL only accessible with enzymes!
  - ✓ Material assessment to position its possible application (biomedical fibers)

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## Acknowledgements

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Dr. ir. A.R.A. Palmans

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Laboratory of Polymer Chemistry  
Laboratory of Polymer Technology  
Laboratory of Macromolecular and Organic Chemistry  
Laboratory of Process Design

Number of publications generated from this project: 8

### **Collaborators in:**

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Hogeschool Nijmegen, HAN  
DSM Research  
KU Leuven  
University of Nottingham  
CIDETEC, San Sebastian

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